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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.006 Å R factor = 0.070 wR factor = 0.165 Data-to-parameter ratio = 10.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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2,4-Bis(methoxymethyl)-1,3,5-trimethylbenzene

The title compound, $C_{13}H_{20}O_2$, displays polar ordering owing to intermolecular $C-H\cdots O$ hydrogen bonds that link the molecules to form a three-dimensional network. A twofold rotation axis bisects the ring, passing through the unsubstituted C atom, the *para*-C atom and C of the attached methyl group.

Comment

In our studies of complexes generated from ditopic ligands (Dobrzańska, 2005; Dobrzańska, Lloyd *et al.*, 2005; Dobrzańska, Raubenheimer *et al.*, 2005; Dobrzańska *et al.*, 2006; Dobrzańska & Lloyd, 2006), we have unexpectedly obtained single crystals of the title compound, (I).



The asymmetric unit of (I) (Fig. 1) consists of half a molecule. The two methoxy groups adopt the trans configuration with respect to the plane of the benzene ring, allowing the molecule to straddle a twofold rotation axis along the c axis. The twofold axis passes through C4, C1 and C5. As a result, the methyl group on the twofold axis is disordered. The C1-C2-C7-O8 torsion angle, which defines the rotation of the H₂C-O bonds out of the plane of the central benzene ring, is -74.5 (4)°. 2,4-Bis(methoxymethyl)-1,3,5-trimethylbenzene molecules pack parallel to the bc plane and display polar ordering (Fig. 2), most likely due to the formation of intermolecular C7–H7A···O8ⁱ hydrogen bonds [symmetry code: (i) $x - \frac{1}{4}, \frac{1}{4} - y, z - \frac{1}{4}$] (Fig. 3 and Table 1). A similar influence of hydrogen bonding on ordering of molecules was observed for 2,6-lutidine, where the presence of linear C-H···N interactions results in macroscopically polar crystals (Bond et al., 2001). Intramolecular C5-H5B···O8 hydrogen bonds are also formed (Table 1). The aromatic rings are at an angle of 17.43° relative to the *bc* plane. The molecules are stacked in a herringbone-like pattern as a result of forming a threedimensional hydrogen-bonded network. π - π Interactions between the benzene rings in adjacent layers is not possible because the aromatic groups are tilted at an angle of 34.86° (centroid-to-centroid) relative to one another. Since the centroid-to-centroid distance between alternate layers is 8.186 Å and the offset is complete, $\pi - \pi$ stacking is implausible in this instance as well.

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Figure 1

The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Unlabeled atoms are related to labeled atoms by the symmetry operation -x, -y, z. Both methyl group disorder components are shown.

Experimental

Colorless crystals suitable for single-crystal X-ray diffraction were obtained by slow evaporation of the solution after an SN_2 reaction of 2,4-bis-(chloromethyl)-1,3,5-trimethylbenzol (0.7238 g) with 4,5-dichloroimidazole in methanol (molar ratio: 0.0033:0.0233).

Z = 8

 $D_x = 1.166 \text{ Mg m}^{-3}$ Mo *K* α radiation

 $\mu = 0.08 \text{ mm}^{-1}$

T = 100 (2) K

 $R_{\rm int}=0.084$

 $\theta_{\rm max} = 28.3^\circ$

Block, colorless

 $0.50 \times 0.38 \times 0.26 \ \text{mm}$

747 independent reflections

623 reflections with $I > 2\sigma(I)$

 $(0.0648P)^2$ $r^2 + 2F_c^2)/3$

Crystal data

 $\begin{array}{l} C_{13}H_{20}O_2\\ M_r = 208.29\\ Orthorhombic, Fdd2\\ a = 14.253 \ (16) \ \text{\AA}\\ b = 20.66 \ (2) \ \text{\AA}\\ c = 8.057 \ (9) \ \text{\AA}\\ V = 2373 \ (4) \ \text{\AA}^3 \end{array}$

Data collection

Bruker APEX CCD area-detector diffractometer ω scans Absorption correction: none 3575 measured reflections

Refinement

$w = 1/[\sigma^2(F_0^2) +$
+ 6.8504P]
where $P = (F_{i})$
$(\Delta/\sigma)_{\rm max} = 0.008$
$\Delta \rho_{\rm max} = 0.32 \ {\rm e}$
$\Delta \rho_{\min} = -0.26 \epsilon$

Tal	ble	1
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		, <u>e</u>		
Hydrogen-bond	geometry	(A, '	°),	•

$D - \mathbf{H} \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
C5−H5 <i>B</i> ···O8	0.98	2.45	3.172 (5)	131
$C7-H7A\cdots O8^{i}$	0.98	2.57	3.542 (7)	166

Symmetry code: (i) $x - \frac{1}{4}, -y + \frac{1}{4}, z - \frac{1}{4}$.

H atoms were positioned geometrically, with C–H = 0.95, 0.99 and 0.98 Å for aromatic, methylene and methyl H, respectively, and constrained to ride on their parent atoms, with $U_{iso}(H) = xU_{eq}(C)$, where x = 1.5 for methyl H, and x = 1.2 for all other H atoms. The





Polar ordering of molecules in the bc plane of the unit cell.



Figure 3

The hydrogen bond pattern (dashed red lines), as viewed along the c axis.

methyl group C5 has its H atoms disordered over two sets of sites, with 0.5 occupancy. In the absence of significant anomalous scattering effects, Friedel pairs were merged.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour 2001); software used to prepare material for publication: *SHELXL97*.

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